

Question Number	Answer	Acceptable answers	Mark
<b>1(a)</b>	A		<b>(1)</b>

Question Number	Answer	Acceptable answers	Mark
<b>1(b)(i)</b>	both points correctly plotted (1)	allow +/- half square	<b>(1)</b>

Question Number	Answer	Acceptable answers	Mark
<b>1(b)(ii)</b>	smooth curve (1) ( does not need to go through all points i.e. can miss out top section)	allow slight discontinuities/double lines/ thick lines  NOT dot to dot /two straight lines	<b>(1)</b>

Question Number	Answer	Acceptable answers	Mark
<b>1(b)(iii)</b>	temperature from 34 °C to 39 °C inclusive (1)		<b>(1)</b>

Question Number	Answer	Acceptable answers	Mark
<b>1(b)(iv)</b>	21(°C) (1)	22( °C ) /23(°C )	<b>(1)</b>

Question Number	Answer	Acceptable answers	Mark
<b>1(c)(i)</b>	it/black is a good absorber of heat /energy/radiation/IR (1) i.e. it absorbs/takes in more infrared/IR	<b>allow</b> it/black absorbs/takes in heat  <b>ignore</b> attracts/emitter/conductor <b>NOT</b> (so it ) cools down quickly	<b>(1)</b>

Question Number	Answer	Acceptable answers	Mark
<b>1(c)(ii)</b>	substitution (1) $9000 \div 20$  evaluation (1)  450 (W)	ignore powers of 10 until evaluation  e.g. $90 \div 2$ gains 1 mark $45$ gains 1 mark  give full marks for correct answer, no working	<b>(2)</b>

Question Number	Answer	Acceptable answers	Mark
<b>1(c)(iii)</b>	substitution (1) $9000 \div 18\ 000$ ( x 100% )  evaluation (1)  50 (%)	ignore powers of 10 until evaluation  e.g. $90\ 000 \div 1800$ gains 1 mark $5$ gains 1 mark  0.5 or $\frac{1}{2}$ or half gains both marks  give full marks for correct answer, no working	<b>(2)</b>

Question Number	Answer	Acceptable answers	Mark
<b>2(a)</b>	A		<b>(1)</b>

Question Number	Answer	Acceptable answers	Mark
<b>2(b)(i)</b>	6%	100 - 94	<b>(1)</b>
<b>(ii)</b>	comparing reflected amount for water with any one of the others (1)	saying one {named material (on the graph) is/all materials (on the graph) are} solid	<b>(1)</b>

Question Number	Answer	Acceptable answers	Mark
<b>2(c)(i)</b>	An explanation to include the following <ul style="list-style-type: none"> <li>• more thermal (heat) energy is absorbed (1)</li> <li>• because water (liquid) absorbs more than ice (solid) (1)</li> </ul>	<p>more radiation is absorbed</p> <p>because water (liquid) reflects less than ice (solid)</p> <p>because less ice surface to reflect</p> <p>because more water surface to absorb</p>	<b>(2)</b>

Question Number	Answer	Acceptable answers	Mark
<b>2(c)(ii)</b>	its temperature rises	<p>gets hotter</p> <p>water level increases/gets higher</p> <p>Ignore '{water/it} {increases/rises}'</p> <p><b>Reject</b> toxicity</p>	<b>(1)</b>

Question Number		Indicative content	Mark
<b>QWC</b>	<b>*2(d)</b>	<p>A description including some of the following</p> <ul style="list-style-type: none"> <li>• solar / heat / light</li> <li>• photosynthesis</li> <li>• chemical / fossil fuel</li> <li>• burning</li> <li>• thermal</li> <li>• in steam</li> <li>• kinetic</li> <li>• in turbine</li> <li>• electrical</li> <li>• in generator</li> </ul>	<b>(6)</b>
<b>Level</b>	<b>0</b>	no rewardable material	
<b>1</b>	<b>1-2</b>	<ul style="list-style-type: none"> <li>• a limited description which identifies an energy in an appropriate place e.g. thermal energy in the boiler OR e.g. the (same) energy flows from the boiler to the turbine</li> <li>• the answer communicates ideas using simple language and uses limited scientific terminology</li> <li>• spelling, punctuation and grammar are used with limited accuracy</li> </ul>	
<b>2</b>	<b>3-4</b>	<ul style="list-style-type: none"> <li>• a simple description which includes details of a relevant energy transfer e.g. (steam causing) the turbine to rotate turns the coil in the generator transferring kinetic energy into electrical energy</li> <li>• the answer communicates ideas showing some evidence of clarity and organisation and uses scientific terminology appropriately</li> <li>• spelling, punctuation and grammar are used with some accuracy</li> </ul>	
<b>3</b>	<b>5 - 6</b>	<ul style="list-style-type: none"> <li>• a detailed description to includes details of a sequence of transfers e.g. chemical energy stored in the coal is transferred in the boiler to thermal energy producing steam. The steam turns the turbine which turns the coil.</li> <li>• the answer communicates ideas clearly and coherently uses a range of scientific terminology accurately</li> <li>• spelling, punctuation and grammar are used with few errors</li> </ul>	